

THE TOTAL SYNTHESIS OF ( $\pm$ )-FRIEDELIN, AN UNSYMMETRICAL, PENTACYCLIC TRITERPENE

Robert E. Ireland\* and David M. Walba<sup>1</sup>

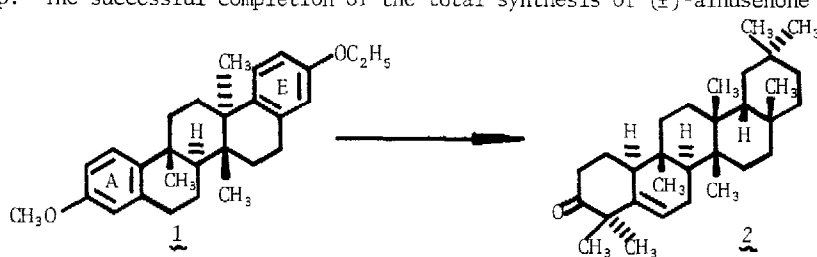
Division of Chemistry and Chemical Engineering

California Institute of Technology, Pasadena, CA 91125

Contribution No. 5261

(Received in USA 20 January 1976; received in UK for publication 27 February 1976)

In previous reports<sup>2</sup> from these laboratories an efficient route to the pentacyclic diether 1 was described. A pentacyclic diaromatic compound of this type, with the trans-anti-trans BCD ring structure and the correct array of angular methyl groups, was envisaged as a useful key intermediate for the total synthesis of pentacyclic triterpenes of the friedelin (10)<sup>3</sup> structural group. The successful completion of the total synthesis of ( $\pm$ )-alnutenone (2)<sup>2</sup> established the

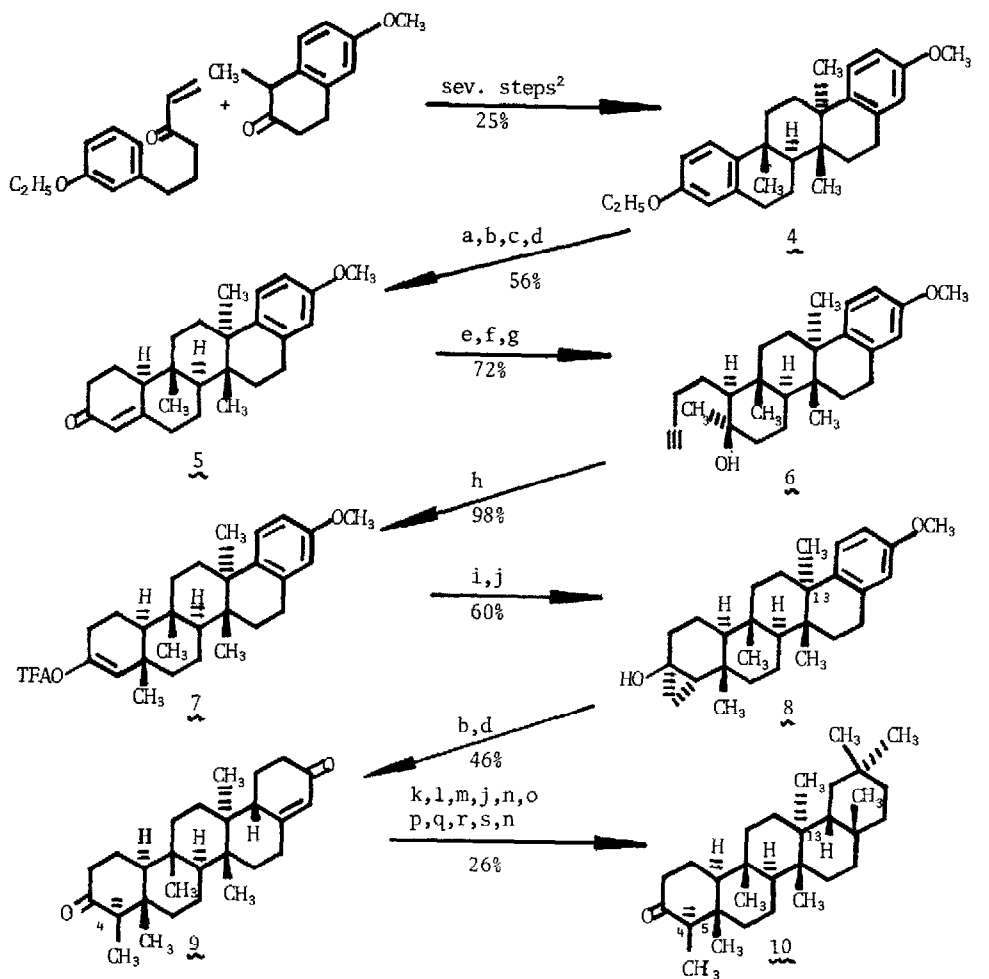


viability of this approach, and the present report describes the results of a continuation of this effort which has culminated in the first total synthesis of ( $\pm$ )-friedelin (10) itself.

Certain modifications of the previous ( $\pm$ )-alnutenone (2) synthesis were deemed wise at the outset of the present work. In the alnutenone scheme<sup>2</sup> the aromatic E ring of the diether 1 was reduced and methylated before reduction and methylation of the aromatic A ring. The known<sup>2,4</sup> ease of the latter transformation dictated this strategy. For the friedelin (10) work such a sequence has disadvantages. First, the overall yield observed<sup>2</sup> for the alteration of the aromatic E ring was only 14%, principally as a result of the initial Birch reduction. It was felt that this could be improved in the friedelin scheme by reversing the aromatic ring modification sequence--i.e., A before E rather than E before A.

Secondly, the process envisaged for the alteration of the aromatic A ring entailed a cationic cyclization as one step (eq. 6 $\rightarrow$ 7). Had this reaction been performed on the acetylenic alcohol precursor 3 in which the severe steric strain in the CDE ring system had already been introduced by prior modification of the aromatic E ring, there was a distinct possibility that backbone rearrangement<sup>3</sup> would compete effectively with cyclization. Thus, it seemed wise to undertake modification of the aromatic A ring first; completion might then lead to a higher yield in the

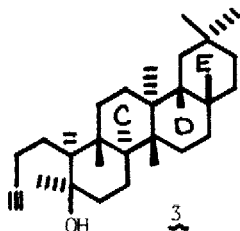
SCHEME I: TOTAL SYNTHESIS OF (±)-FRIEDELIN (10)<sup>a</sup>



(31 steps, 0.3% overall yield)

<sup>a</sup> a, (C<sub>6</sub>H<sub>5</sub>)<sub>2</sub>PLi, THF<sup>5</sup>; b, Li, NH<sub>3</sub>, DME, EtOH; c, CH<sub>3</sub>I, DMF; d, 5NHCl, EtOH, C<sub>6</sub>H<sub>6</sub>; e, H<sub>2</sub>O<sub>2</sub>, aq. NaOH, CH<sub>3</sub>OH; f, pTsNHNH<sub>2</sub>, HOAc, CH<sub>2</sub>Cl<sub>2</sub><sup>6</sup>; g, CH<sub>3</sub>Li, Et<sub>2</sub>O; h, CF<sub>3</sub>CO<sub>2</sub>H, (CF<sub>3</sub>CO)<sub>2</sub>O; i, LDA, THF; j, Zn-Ag, CH<sub>2</sub>I<sub>2</sub>, THF<sup>7</sup>; k, Li (OtBu)<sub>3</sub>AlH, THF, C<sub>6</sub>H<sub>6</sub>, O<sup>9</sup>; l, DHP, POCl<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>; m, Li (OtBu)<sub>3</sub>AlH, THF, C<sub>6</sub>H<sub>6</sub>, reflux; n, CrO<sub>3</sub>·2Py, CH<sub>2</sub>Cl<sub>2</sub>; o, KOtBu, CH<sub>3</sub>I, THF; p, Li, NH<sub>3</sub>, THF, tBuOH; q, ClPO(NMe<sub>2</sub>)<sub>2</sub>, DME, HMPA, nBuLi<sup>8</sup>; r, Li, EtNH<sub>2</sub>, tBuOH; s, pTsOH, CH<sub>3</sub>OH, THF.

Birch reduction but, more important, would relegate the introduction of the principal source of steric strain--the cis-D/E ring fusion--to the later, less sensitive stages.



To implement this scheme, it was necessary to prepare the pentacyclic diether 4 in which the location of the ethoxy and methoxy groups were interchanged from their locations in the ( $\pm$ )-alnutenone precursor 1 (SCHEME I). This was accomplished in 25% overall yield from the appropriate precursors shown by the same procedures previously reported.<sup>2</sup>

Selective reduction of the aromatic A ring did occur in superior yield in this case, and modification of this ring along the same lines as reported<sup>9</sup> previously in the total synthesis of ( $\pm$ )-shionone led through the acetylenic alcohol 6 to the enol trifluoroacetate 7. Methylenation of the enolate derived from this material 7 by treatment with lithium diisopropylamide provided the cyclopropyl alcohol 8.

In contrast to the experience in the ( $\pm$ )-shionone synthesis,<sup>9</sup> cleavage of the cyclopropyl alcohol 8 was not particularly efficient at this state. While both base (KOTBu, Et<sub>2</sub>O) and acid (5NHC1, EtOH, C<sub>6</sub>H<sub>6</sub>) afforded predominately the desired 4-methylketone, in neither case was the yield better than 75%, and when coupled with the subsequent Birch reduction of the aromatic E ring, the enedione 9 was available only in 35% overall yield. Since the complication in the more favorable acid catalyzed cleavage appeared to entail backbone rearrangement<sup>3</sup> of the pentacyclic system not seen in the ( $\pm$ )-shionone work, an adequate solution to the problem was found when the aromatic E ring was reduced first and then the resulting dihydrocyclopropyl alcohol cleaved by acid treatment.

Completion of the synthesis from the enedione 9 by the introduction of the requisite E ring methyl groups followed the pattern used in the previous ( $\pm$ )-alnutenone synthesis<sup>2</sup> after provision was made to protect the C3 oxygen function. ( $\pm$ )-friedelin (10), mp 246.5-248<sup>0</sup>, so formed gave tlc and glpc behavior and infrared, pmr and mass spectra identical to those of an authentic sample of the natural product isolated from cork.<sup>3</sup>

#### Acknowledgment

Support of this work by grants from the National Science Foundation and the Hoffmann-LaRoche Foundation is gratefully acknowledged.

References

- 1) National Defense Education Act Trainee, 1971, 1974.
- 2) R. E. Ireland, M. I. Dawson, S. C. Welch, A. Hagenbach, J. Bordner and B. Trus, J. Amer. Chem. Soc., 95, 7829 (1973).
- 3) E. J. Corey and J. J. Ursprung, J. Amer. Chem. Soc., 78, 5041 (1956); G. Brownlie, F. S. Spring, R. Stevenson and W. S. Strachen, J. Chem. Soc., 2419 (1956).
- 4) R. E. Ireland and L. N. Mander, J. Org. Chem., 32, 689 (1967).
- 5) F. G. Mann and M. J. Pragnell, Chem. Ind. (London), 1386 (1964).
- 6) D. Felix, J. Schreiber, G. Ohloff and A. Eschenmoser, Helv. Chim. Acta, 54, 2896 (1971).
- 7) H. W. Whitlock, Jr. and L. E. Overman, J. Org. Chem., 34, 1962 (1969); J. M. Denis, C. Girard and J. M. Conia, Synthesis, 549 (1972).
- 8) R. E. Ireland, D. C. Muchmore and U. Hengartner, J. Amer. Chem. Soc., 94, 5098 (1972).
- 9) R. E. Ireland, C. A. Lipinski, C. J. Kowalski, J. W. Tilley and D. M. Walba, J. Amer. Chem. Soc., 96, 3333 (1974).